SYNTHESIS AND EVALUATION OF 3-CARBOXYMETHYL STEROIDS AS INHIBITORS OF HUMAN PROSTATIC STEROID 5\(\alpha\)-REDUCTASE

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Abstract: A series of 3-carboxymethyl steroids has been prepared and assayed *in vitro* as inhibitors of steroid 5α -reductase isolated from human prostatic tissue. Potent inhibition was observed with analogs from both androstene and estratriene families. Phosphonic and sulfonic acid analogs were also prepared and found to lack significant inhibitory activity.

Dihydrotestosterone (DHT), a metabolite of the major circulating androgen testosterone, produced by the NAPDH-dependent-enzyme steroid 5α -reductase, is recognized to be the principal mediator of several androgen-related conditions. The most notable of these is benign prostatic hyperplasia (BPH), an ailment afflicting a remarkably large percentage of aging males, which is often accompanied by the discomforting symptoms of urinary dysfunction. In addition, elevated levels of DHT have been implicated in the etiology of some prostatic cancers and several skin disorders including acne, male pattern baldness, and hirsutism. As a result of the elucidation of these physiological roles of DHT within the past twenty years, an active search has ensued for potent inhibitors of steroid 5α -reductase as potential therapeutics. In 1992 the first such inhibitor (finasteride was approved in the U.S. for treatment of BPH. While the steady state kinetic mechanism of steroid 5α -reductase has been well characterized, only recently has molecular characterization revealed the existence of two isozymes encoded by distinct genes. Elucidating the tissue expression and physiological roles of these enzyme isoforms remain areas of intense research.

As part of our ongoing effort to discover novel inhibitors with potential pharmaceutical utility, we have previously reported on the synthesis and characterization of several classes of potent inhibitors of human prostatic steroid 5α -reductase activity exemplified by compounds 1, 2, and 3.9-14

The first of these compound classes (steroidal acrylates, e.g. 1 and 2) derived from consideration of the chemical mechanism of the enzyme and were designed as potential mimics of the putative enolate intermediate -- the key feature being the use of a carboxylic acid functional group as a stable, anionic surrogate for the enolate oxyanion.^{9,10} Subsequently, we demonstrated inhibitory activity by the analogous estrone-derived A-ring aromatic carboxylic acids (e.g. 3)¹¹ and later by phosphonic,¹² phosphinic,¹² and sulfonic acid analogs.¹³ We proposed that the anionic acid functionality coordinates to an active site electrophile(s) which may be present to activate the substrate enone for hydride transfer, to stabilize the resulting enolate, and/or to transfer a proton to the steroidal substrate.¹⁵

The notion that the C-3 carboxylate coordinates to an enzyme active site electrophile led us to synthesize homologated analogs of 1, 2 and 3 which, through greater conformational freedom of the C-3 appendage, might allow for an enhanced interaction of this nucleophilic group with active site electrophiles.

Herein we report the synthesis and evaluation of several homologated analogs which exhibit a range of inhibitory potencies against human prostatic steroid 5α-reductase activity.

Chemistry. The synthesis of 4 (Table 1) was accomplished in two steps (Scheme 1) from the known enol triflate 16 or relying on a palladium-catalyzed cross-coupling of a Reformatsky reagent. The isomeric Δ^2 - β,γ -unsaturated acid 5 was prepared by reacting a Horner-Emmons reagent under equilibrating conditions (NaOEt, EtOH) with ketone 17, whereas treatment of 17 with the same phosphonate reagent and NaH in DMF provided, after hydrolysis, the α,β -unsaturated acid 9. Horner-Emmons reaction using enone 18 afforded a mixture of (\mathcal{E}) - $\alpha,\beta,\gamma,\delta$ -unsaturated dienyl and $\Delta^{3,5}$ - $\beta,\gamma,\delta,\varepsilon$ -unsaturated esters which were separable after hydrolysis to yield 6 and 7. Addition of *in situ*-generated lithium methoxyacetylide one 18 followed by acid treatment and ester hydrolysis led to the isolation of the isomeric (\mathcal{Z})-conjugated dienyl acid 8 as a separable mixture with 7.

Scheme 1.19

a) (Ph₃P)₂PdCl₂, DIBAL, 0°, then BrZnCH₂CO₂^tBu, r.t., b) LiOH, H₂O, MeOH, THF, reflux; c) (EtO)₂P(O)CH₂CO₂Me, NaOEt, EtOH, reflux; d) K₂CO₃, MeOH, H₂O, reflux; e) (EtO)₂P(O)CH₂CO₂Me, NaH, DMF, reflux; f) Br₂HCCHClOMe, nBuLi, -78° to 0°; H₂SO₄; g) allylSnBu₃, (Ph₃P)₂PdCl₂, LiCl, dioxane; h) O₃; Me₂S; i) Jones reagent, j) 9-BBN; H₂O₂, KOH; k) SOCl₂; l) NaCN, Bu₄NBr; m) conc. HCl; n) CH₂N₂, o) DAST; p) NaBH₄, q) P(OEt)₃; r) TMSCl, Nal; s) PBr₃; t) (H₂N)₂CS, NaOH; u) Cl₂, HOAc; K₂CO₃, H₂O, acetone.

One and two carbon homologated A-ring aryl acids 10 and 11 were prepared from aryl triflate 19 via the intermediacy of the allylstannane coupling product $20.^{18}$ Ozonolysis of 20 followed by Jones oxidation led to homolog 10 while hydroboration/oxidation led to the double homolog 11. The α -keto and α , α -difluoro analogs of 10 were synthesized from the previously described 11 aryl acid 3. Formation of the acyl cyanide followed by acid hydrolysis afforded the ketoacid 12. Treatment of the methyl ester of 12 with diethylaminosulfur trifluoride (DAST) and subsequent hydrolysis provided difluoro acid 13. Finally, preparation of the phosphonic and sulfonic acid analogs proceeded through the 3-derived benzyl alcohol 21. Arbuzov reaction followed by dealkylation with TMSI provided the benzyl phosphonic acid analog 14. Conversion of alcohol 21 to the benzyl bromide followed by thiourea displacement, hydrolysis, and oxidation provided the corresponding sulfonic acid analog 15.

Enzyme Inhibition. In spite of the greater steric demands of the C-3 appendage, the regioisomeric acrylate homologs 4 and 5 exhibited potent inhibition of human prostatic steroid 5α -reductase activity²⁰ (inhibition constants of 85 and 40 nM, respectively, Table 1), comparable to the parent acrylate 1 ($K_{i,app}=30$ nM) and its Δ^2 isomer (not shown, $K_{i,app}=85$ nM).¹⁰ In contrast, the $\Delta^{3,5}$ dienyl acid 6 was a relatively weak inhibitor ($K_{i,app}=450$ nM) as compared to its parent 2 ($K_{i,app}=7$ nM).¹⁰ While the original design rationale

was to provide greater conformational freedom for the carboxylate in the active site, compounds 7, 8, and 9 with exocyclic double bond configurations were readily accessible and allowed examination of two different but specific orientations of the carboxylate. However, these compounds were devoid of measurable inhibitory activity.

Table 1.

Compd No.	Structure	K _{i,app} (nM)	Compd No.	Structure	K _{i,app} (nM)
4	HOSC H	2 85	10 но ₂ 0	CON(iPr);	20
5	HO ₂ C H	40	11 но <i>⊊</i> с	CON(iF	290
6	HO ₂ C. CON(iPr) ₂	450	12 но _г о	CON(iPr)2	150
7	HO ₂ C CON(iPr) ₂	>5000	13 HO ₂ Ω	CON(iPr)	40
8	CON(iPr) ₂	>5000	14	CON((IPr)	740
9	HO ₂ C A CON(iPt) ₂	>5000	15 но _{\$} с	CON(iPr)	>5000

Homologs in the estrane series also demonstrated potent inhibition as exemplified by 10 which, with a $K_{i,app}$ = 20 nM, is equipotent with its parent, aryl carboxylate 3.11 Extending the connecting chain to two carbons (compound 11) resulted in an inhibitory activity reduced by an order of magnitude. Substitution of the benzylic carbon with gem-difluoro and oxo groups (which not only lower the carboxylate pK_a but also act as potential hydrogen bond acceptors) resulted in a maintenance of activity in the former (13, $K_{i,app}$ = 40 nM) and a diminution of activity in the latter case (12, $K_{i,app}$ = 150 nM). Finally, extending this inhibitor design strategy to phosphonic and sulfonic acid analogs 14 and 15 resulted in a remarkable loss of activity ($K_{i,app}$ = 740 and >5000 nM, respectively) compared to their non-homologated parents 12,13 which were characterized by inhibition constants in the range of 20 to 50 nM.

In summary, inhibition of human prostatic steroid 5α -reductase has been demonstrated with a series of 3-carboxymethyl steroids. These compounds derive from previously described 3-carboxy steroid classes of inhibitors yet structure-activity relationships within the parent and homologated series do not entirely correspond.

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References and Notes

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- All new compounds were characterized by ¹H-NMR. Compounds **4-15** provided satisfactory elemental analyses and mass spectra. Melting points: **4**: 236-239°C, **5**: 270°C, **6**: 205-207°C, **7**: 308°C, **8**: 215-217°C, **9**. foam, **10**: 125-127°C, **11**: 110-115°C, **12**: 251-253°C, **13**. 222-224°C, **14**: 190°C, **15**: 275°C (as potassium salt).

Regiochemical assignments for isomeric 4 and 5 are consistent with the maintenance of double bond position in the Pd-catalyzed coupling of 16 and the isomerization of the double bond of 5 into the thermodynamically more stable position under equilibrating conditions. The downfield chemical shift for the vinylic proton of 5 relative to 4 is consistent with that observed in the analogous acrylates (e.g. $1)^{10}$. Partial 1 H-NMR for 4: δ 2.95 (s, 2H, HO₂C-CH₂-), 5.2 (s, 1H, H-4); 5' δ 3.0 (s, 2H, HO₂C-CH₂-), 5.5 (d, J = 5.6 Hz, 1H, H-2). Structural assignment of 6 was consistent with nOe observed for H-6 and H₂-21 (HO₂C-CH₂-) upon irradiation of H-4. Partial 1 H-NMR for 6: δ 3.06 (s, 2H, HO₂C-CH₂-), 5.40 (d, J = 3.0 Hz, 1H, H-6), 5.87 (s, 1H, H-4). Geometrical isomers 7 and 8 were assigned as E and E respectively, based on observed nOe between H-4 and H-21 of 7 and no such enhancements for 8. The downfield shift for H-4 of 8 relative to that of 7 is consistent with a deshielding effect of the proximal carboxylate in 8. Partial 1 H-NMR for 7: δ 5.54 (s, 1H, H-4), 5.84 (s, 1H, H-21), 8: δ 7.18 (s, 1H, H-4), 5.41 (s, 1H, H-21). Geometrical assignment for 9 was based on an observed nOe for H-4a upon irradiation of H-21 H-4a and H-4b were assigned by two-dimensional 1 H-NMR (COSY) correlations. Compound 9 partial 1 H-NMR δ 1.93 (H-4a), 2.22 (H-4b), 5.62 (s, H-21).

- Evaluation of steroid 5α-reductase activity was performed with microsomal human prostatic enzyme preparations as
 previously described.⁹ Apparent inhibition constants (K_{I,app}) of test compounds were determined by the method of
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